



LECTURE SCHEDULE

Training Course

ORGANOMETALLIC CATALYSIS

Professor Jean Pierre Genet (ParisTech-Ecole Nationale Supérieure de Chimie)

On November 27th, 28th, 29th, 30th, 2017

ESPCI 10 Rue Vauquelin 75005 Paris France

Day 1

9h00-10h30	 Lecture 1: Basic chemistry of transition metals complexes Introduction Historical reactions on homogeneous catalysis: few main reactions Formation of transition metal complexes Electronic configurations. The 16 and 18 electrons rule Definition and interest of organometallic catalyst: MtLn General presentation of organic ligands and nomenclature Electrons counting and degree of oxidation Examples and applications Exercises on organometallic chemistry: Problems set
10h30-10h45	Coffee break
10h45-12h15	 Lecture 2: Elementary steps in organometallic chemistry Introduction Non-reductive elimination Oxidative addition Insertion reaction Reductive elimination Transmetallation Ligands substitution Coordination Insertion Nucleophilic addition on ligand coordinated to transition metal β and α elimination of hydrogen atoms (dehydrometallation) Applications: mechanisms of Stille, Suzuki, Heck, Sonogashira coupling reactions.
12h15-14h00	Lunch
14h00-15h30 Main types of ligands	Lecture 3: Notions de ligands - Alkyls, allyls . Phosphines, water-soluble phosphines . Phosphites - Phosphoramidites - Phospholes - Amines - Bisoxazolines and imidazolines - P-N ligands: phosphinooxazoline - Carbenes - Carbon monoxide
15h30-15h45	Coffee break
15h45-17h15	Lecture 4: Reactivity of metal complexes

	 Metal ligand interactions (σ et π effects) Factors influencing the reactivity and the selectivity Role of ligands Steric parameters (Tolman cone angle) Electronic effects Example: optimization of L-DOPA synthesis
Day 2	
9h00-10h30	Lecture 5: Double bond hydrogenation reactions
	-Introduction -Preparation of complexes (Rh,Ir)
Hydrogenations of non-s	substituted olefins (Rh,Ir) -Osborn-Wilkinson catalyst -Mechanism -Crabtree catalyst: synthesis and applications Selectivity (hydrogeneticate assisted by functional groups
Asymmetric hydrogenat	 Selectivity/ hydrogenations assisted by functional groups ion of prochiral olefins Chiral ligand Dehydroaminoacids synthesis Application to the synthesis of L-DOPA Origin of enantioselectivity
Asymmetric hydrogenat (API): thrombin inhibito endopeptidase C inhibito succinate moiety of meta anticonvulsant, building vitronectin receptor anta	 ion of alkenes (Ru) Preparations of ruthenium chiral complexes Allylic alcohols Applications in perfumery (lilial, florydral) α,β-Unsaturated acids and dehydroaminoacids Mechanism Application to the antiflamatory drug: Naproxen Enantioselectivitity orientation : comparison between Ru (II) and Rh (I) catalysts Asymmetric hydrogenation of non-functionalized olefins, lactones and unsaturated ketones Application in perfumery: an industrial example the Paradisone Applications to the synthesis of active pharmaceutical ingredients r CRC 220, synthesis of a renin inhibitor , anti-HIV drug indinavir, or candoxatril, endothelin antagonists such as SB 209670, chiral allo- protease inhibitor UK-370,106, pregabalin a marketed blocks for a new prototype of HCV protease inhibitor BILN 2061, gonist SB-273005 Examples of catalysts recovery
10h30-10h45	Coffee break
10h45-12h15	Lecture 6: Hydrogenation of carbonyls and imines

- Preparation of catalysts

	- Hydrogenation of ketoesters
	- Mechanism
	- Enantioselectivity-offentation - Selectivity: influence of the dihedral angle of the metal-
	diphosphine catalyst
	- Applications in organic synthesis and comparison with
	biochemical reduction
	- Hydrogenation of fluorinated β -diketones (optimization of
	enantioselectivity with electron deficient diphosphine)
	- Application to the synthesis of rivastatin
	- Dynamic kinetic resolution (DKR
	- Application: synthesis of a key chiral intermediate of the
	- Synthetic applications
Reductions by hydrogen	transfer (Rh Ru Fe)
iteauetions of nyarogen	- History
	- The different developments
	- Noyori reaction: mechanism and applications
	- Applications for the synthesis of pharmaceutical intermediates:
	NMDA 2B Receptor Antagonist Ro 67-8867, synthesis of
	neurokinin receptor antagonist, vasopressin receptor antagonist
	OPC-4106, enantioselective approach to PDE-IV inhibitor
Asymmetric hydrogenati	on of imines
r is j innetite it j ut o genuti	- Catalysts
	- Use in organic synthesis: chiral synthesis of the antibiotic (S) -
	levofloxacin, synthesis of the neuromuscular blocker GW
0430,	AMPA receptor modulator S-1898
	- Industrial synthesis of Metolaclor
	- Conclusions
12h15-14h00	Lunch
14h00-15h30	Lecture 7: Isomerizations
Isomerizations of double	bonds
	- Mechanism: 1,2 and 1,3 hydrogen shifts
	- Isomerization of allylic and propargylic alcohols
	- Isomerization of ethers
	- Applications in perfumery
	- Kinetic resolution
	- Isomerization of allylic acetates Machanisms: comparison between Pd (II) and Pd (0) cotalysts
	- Chirality transfer
Enones isomerization	
	Catalysts
	Mechanisms
	Examples
15h30-15h45	Coffee break

15h45-17h15 Lecture 8: Asymmetric isomerization

Isomerization of allyl amines.

9h00-10h30 Lecture 9 : Oxidations I

-	- Catalysts (Mo, Rh, Ir)
	- Chemoselectivity : Comparison between isomerization catalyzed
	by bases/acids and isomerization catalyzed by transition metals
	- Enantioselectivity
	- Mechanism
	- Developments in organic synthesis
	- Industrial applications: synthesis of menthol, ambrox
	- Enantioselective isomerization of allylic alcohols
	- Applications in perfumery
	- Synthesis of florydral
Other substrates	
	- Isomerization of allylic and propargylic alcohols
	- Applications for prenal, geraniol and citral synthesis
	- Epoxides isomerization (Pd)
	- Secondary alcohols racemization (Ru)
	- Dynamic kinetic resolution (DKR): Chemoenzymatic reactions

Day 3

Wacker oxidation (Pd)	
	- Mechanism
	- Industrial applications
Olefins epoxidation	
	- Different classes of oxidizing agents
	- Oxidations catalyzed by transition metals
	- Chemoselectivity
	- Asymmetric epoxidation of allylic alcohols
	- Sharpless epoxidation
	- Origin of enantioselectivity
	- Kinetic resolution
	- Synthetic applications
Other methods	
	Chiral disaring as
	- Chiral dioxiranes Katsuki Jacobsen reaction
	- Origin of enantioselectivity
	- Mechanism
	- Synthetic application: taxotere side chain
10h30-10h45	Coffee break
10h45-12h15	Lecture 10: Oxidations reactions II

Dihydroxylation of olefins (AD)

- History
- Chiral ligands
- Mechanism
- Synthetic applications

Oxidation of sulfur compounds

- Catalytic systems
- Chiral sulfoxides

- Applications to large scales production of sulfoxides of industrial interest : Esomeprazole used in the treatment of gastroesophageal reflux disease, , ACAT inhibitor

12h15-14h00 Lunch

14h00-15h30 Lecture 11: Carbonylations

Hydroformylation of olefins

	- Introduction
	- Catalysts and mechanisms
	- Regioselectivity
	- Catalysts and stereo-oriented hydroformylation
	 Hydroformylation of propene use of water-soluble catalysts (Rh-TPPTS) Hydroformylation of alkynes Asymmetric hydroformylation Applications in perfumery
	 Carbonylation of compounds with benzyl and allyl chlorides Double carbonylation of benzyl chlorides Carbonylation of methanol
	- Acetic acid symmesis
15h30-15h45	Pause café
15h45-17h15	Lecture 12: Metathesis
Carbenes	
Curbenes	Introduction
	Structure of carbenes
	Generation of carbenes : Fisher and Schrock types
	Synthesis and functionalization of carbenes
	Reactivity of carbenes
	Tebbe reagent
	Olefination of esters and amides
	Catalytic reactions and applications in organic synthesis
Olefins metathesis	
	- Different types: RCM, ADMET, ROMP
	- Mechanism: Y. Chauvin
	- Industrial processeses

	 Catalysts Mo, Ru Ring formation (RCM), Mechanism Industrial synthetic applications (medium ring, macrolactonisations) Stagonolide A,Decalactone Application to the synthesis of HCV protease inhibitor BILN 2061, Asymmetric synthesis Enynes metathesis
Cross Metathesis (CM)	 Alkenes, Synthesis of Aplysamine Alkynes metathesis: molybdenum and tungsten catalysts Formation of ring from alkynes (RCAM)
Conclusions	- Pormation of Fing Hom arkynes (RCAW)
Day 4	
9h00-10h30	Lecture 13: Creation of bonds C-C (I)
Cyclopropanations Palladium/ Nickel cataly	 Carbenoïdes Mechanism C-H and N-H insertion: inter and intramolecular reactions Applications zed reactions ; C-C and C-N bonds formation Sonogashira reaction (synthesis of galbanolene) Negishi coupling and Heck reactions Application to the syntheses of prosulfuron (herbicide) and pharmaceutical compounds : carbacyclin,discodermolide, xerulinic acid , singulair Suzuki coupling (Pd, Ni) and applications to the synthesis of pharmaceutical drugs: prostaglandins, palytoxin, myxalamide, Valsartan (Novartis), Valsartan (Novartis), ABT-963(Abbot), RO0094889 (Roche) Buchwald-Hartwig reaction Allylic substitution (Tsuji-Trost reaction) Application to the synthesis of the neuromuscular blocker GW 0430, AMPA receptor modulator S-18986, synthesis of MGS0028, an mGluR 2 receptor agonist α arylation of enolates Intramolecular arylation
10h30-10h45	Coffee break
10h45-12h15	Lecture 14: Formation of C-C bonds (II)
Dienes / Applications in	fine chemistry - Telomerizations /applications synthesis of jasmonate - Additions - Cycloadditions

	- Few examples of applications in organic synthesis
Cycloisomerizations	Enynes, DiynesMechanismApplications
12h15-14h00	Lunch
14h00-15h30	Lecture 15: C-H Activation: towards a green catalysis
	- Introduction
	- Olefination of $C(sp^2)$ H Bonds
	 Arylation of C(sp²) H and C(sp³) H bonds- Ortho Alkylation Sequential olefination of aryl iodides
	- Arylation and Alkylation of $C(sp^2)$ H and $C(sp^3)$ H bonds - Conclusions and perspectives
15h30-15h15	Coffee break
15h15-16h45	Lecture 16: Other types of reactions
Asymmetric Michael re	actions
	- Additions of organoboron compounds
	 Boronic acids and organopotassium trifluoroborates to Michael acceptors
	- Stereochemical outcome of the asymmetric 1,4-addition
	Reactions
	- Application to the synthesis of tolterodine
	- Synthesis of chiral α and β -amino acids (Mechanism and
	DFT calculations)
	- Epoxides opening
- Dienes isomerization	ns : Claisen, Cope reactions
D	
- Protecting groups in	fine chemistry

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